### Lesson Plan: Teaching Chemistry of Carbon Compounds through Climate-related Examples

As a high school Chemistry teacher, you can use this set of computer-based tools to help you in teaching the chemistry of carbon and its compounds, the interaction of greenhouse gas molecules with electromagnetic radiation, and environmental chemistry.

This lesson plan allows students to visualize the molecular structure of atmospheric gases and to understand the effect of electromagnetic radiation on these molecules. The activity will also introduce the topic of greenhouse gases and their role in climate change.

Thus, the use of this Lesson Plan allows you to integrate the teaching of a climate science topic with a core topic in Chemistry.

Use this lesson plan to help your students find answers to:

- How do molecules of gases interact with electromagnetic radiation?
- How do atmospheric carbon dioxide (CO<sub>2</sub>) molecules interact with infrared photons?
- What is the greenhouse effect of Earth's atmosphere?
- Could an increase in methane emissions affect the Earth's temperature? Why?

#### About the Lesson Plan

Grade Level	High school
Discipline	Chemistry
Topic(s) in Discipline	Interaction of Molecules with Electromagnetic Radiation, Molecular Vibrations, Molecular Structure of Carbon Compounds (CO <sub>2</sub> , CH <sub>4</sub> ), Greenhouse Gases

Climate Topic	The Greenhouse Effect, Climate and the Atmosphere
Location	Global
Access	Online
Language(s)	This Lesson Plan is available in multiple languages
	Link here: <a href="https://tropicsu.org/resources/lesson-plans/trop-icsu-in-different-languages/">https://tropicsu.org/resources/lesson-plans/trop-icsu-in-different-languages/</a>
Approximate Time Required	100–120 min
1 Content	S
1. Reading (5–10 min)	A reading that provides an overview of the interaction between infrared radiation and molecules of different atmospheric gases.
<b>1</b> Content	A reading that provides an overview of the interaction between infrared radiation and molecules of different atmospheric gases. https://scied.ucar.edu/carbon-dioxide-absorbs-and-re-emits-infrared-radiation
<ol> <li>Reading (5–10 min)</li> <li>Micro-lecture (~8 min)</li> </ol>	A reading that provides an overview of the interaction between infrared radiation and molecules of different atmospheric gases. https://scied.ucar.edu/carbon-dioxide-absorbs-and-re-emits-infrared-radiation A micro-lecture (video) that explains the interaction of molecules such as CO <sub>2</sub> and CH <sub>4</sub> with electromagnetic radiation, and the resulting molecular vibrations leading to the greenhouse effect in the atmosphere.
<ol> <li>Reading (5–10 min)</li> <li>Micro-lecture (~8 min)</li> </ol>	A reading that provides an overview of the interaction between infrared radiation and molecules of different atmospheric gases. https://scied.ucar.edu/carbon-dioxide-absorbs-and-re-emits-infrared-radiation A micro-lecture (video) that explains the interaction of molecules such as CO <sub>2</sub> and CH <sub>4</sub> with electromagnetic radiation, and the resulting molecular vibrations leading to the greenhouse effect in the atmosphere. https://www.coursera.org/lecture/global-warming/greenhouse-gas-physics-SvfZD

#### greenhouse gases in climate change.

https://phet.colorado.edu/en/simulation/greenhouse

4. Suggested questions/assignments for learning evaluation

min)

- How do molecules of gases interact with electromagnetic radiation?
- How do atmospheric carbon dioxide (CO<sub>2</sub>) molecules interact with infrared photons?
- What is the greenhouse effect of Earth's atmosphere?
- Could an increase in methane emissions affect the Earth's temperature? Why?



- 1. Introduce the topic through online reading
- Discuss the sources of carbon and its compounds.
- Discuss the molecular structure of some carbon compounds such as carbon dioxide (CO<sub>2</sub>) and methane (CH<sub>4</sub>).

your preferences and requirements.

Here is a step-by-step guide to using this lesson plan in the

classroom/laboratory. We have suggested these steps as a possible plan of action. You may customize the lesson plan according to

- Introduce the topic of atmosphere and its composition. Name the atmospheric gases (including CO<sub>2</sub> and CH<sub>4</sub>) and their typical percentages.
- Discuss the phenomenon of sunlight striking the Earth and infrared radiation being emitted from the Earth.
- Use online reading material to provide an overview of how CO<sub>2</sub> and other atmospheric gas molecules interact with electromagnetic radiation.

		The reading material is available at: <u>https://scied.ucar.edu/carbon-dioxide-absorbs-and-re-emits-</u> infrared-radiation.
2.	Play a micro-lecture (video)	Now play this micro-lecture (video, approx. 8 min) to explain the interaction of molecules such as $CO_2$ with electromagnetic radiation, and the resulting molecular vibrations that lead to the greenhouse effect in the atmosphere.
		The video micro-lecture from David Archer, the University of Chicago, is available at: <a href="https://www.coursera.org/lecture/global-warming/greenhouse-gas-physics-SvfZD">https://www.coursera.org/lecture/global-warming/greenhouse-gas-physics-SvfZD</a>
3.	Conduct an activity using an interactive visualization	Next, explore this topic in an interactive and engaging manner by using a visualization tool, " <u>The</u> <u>Greenhouse Effect</u> ", from PhET. The tool will help your students to visualize the molecular structure of carbon compounds (CO <sub>2</sub> , CH <sub>4</sub> ),
		explore the effect of electromagnetic radiation on the molecules, and understand the role of greenhouse gases in climate change.
		<ul> <li>Download PhET's tool, "The Greenhouse Effect", from <u>https://phet.colorado.edu/en/simulation/greenhouse</u>.</li> </ul>
		<ul><li>Launch the tool.</li><li>Go to the Photon Absorption pane.</li></ul>
		<ul> <li>For each Atmospheric Gas (select options under Atmospheric Gases)—CH<sub>4</sub>, CO<sub>2</sub>, H<sub>2</sub>O, N<sub>2</sub>, O<sub>2</sub>— observe the molecular structure and the effect of the Infrared Photon on the molecule. The slider on the Light Source can be adjusted to control the rate of emission of photons.</li> </ul>
		• Use the Build Atmosphere option to set the number of molecules of each atmospheric gas.

	Visualize the effect of the Infrared Photons on the molecules in the atmosphere.
	• The absorption of infrared photons by gases such as CO <sub>2</sub> and CH <sub>4</sub> in the atmosphere results in the warming of the surface of the planet. This effect is called the greenhouse effect.
	Go to the Greenhouse Effect pane.
	<ul> <li>Select different scenarios (Atmosphere during), observe the Greenhouse Gas Composition for each scenario, and the corresponding surface temperature shown in the thermometer on the left. Draw inferences.</li> </ul>
4. Questions/Assignments	Use the tools and the concepts learned so far to discuss and determine answers to the following questions:
	How do molecules of gases interact with electromagnetic radiation?
	• How do atmospheric carbon dioxide (CO <sub>2</sub> ) molecules interact with infrared photons?
	• What is the greenhouse effect of Earth's atmosphere?
	• Could an increase in methane emissions affect the Earth's temperature? Why?

# 3 Learning Outcomes

The tools in this lesson plan will enable students to:

- visualize the molecular structure of atmospheric gases
- describe the effect of electromagnetic radiation on these molecules
- identify greenhouse gases and examine their role in climate change

## **4** Additional Resources

If you or your students would like to explore the topic further, these additional resources will be useful.

1. Video	A video in which Iain Stewart demonstrates infrared radiation absorption by CO <sub>2</sub> in a scene from BBC's "Earth: The Climate Wars" documentary:
	https://www.youtube.com/watch?v=kGaV3PiobYk
2. Visualization	An interactive visualization tool, "Greenhouse Gases", from Concord Consortium's Innovative Technology in Science Inquiry:
	https://concord.org/stem-resources/greenhouse-gases

